CLAIMS

1. A process for the preparation of compound AQ4N of formula (2):

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or a salt or solvate thereof wherein the said process includes the reaction step:

where compound AQ4 of formula (1) is oxidised to compound 10 AQ4N of formula (2) with an oxidising agent at a reaction temperature not exceeding $10\,^{\circ}\text{C}$.

- 2. A process according to claim 1 where the oxidising agent in the reaction step is either hydrogen peroxide, an
- 15 oxaziridine, a peracid or a salt of a peracid.

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3. A process according to claim 2 where the oxidising agent is magnesium monoperoxyphthalate.

- 4. A process according to either claim 2 or claim 3 where the reaction is conducted at a temperature not exceeding 0°C.
 - 5. A process according to any one of claims 1 to 4 where the reaction solvent is 1,2-propanediol, dichloromethane or an aliphatic alkyl alcohol.

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6. A process according to any one of claims 1 to 5 for the preparation of a salt of AQ4N, where the salt of AQ4N, or a solvate thereof, is prepared by reaction of compound AQ4N of formula (2) with a solution of hydrogen chloride.

- 7. A process according to any one of claims 1 to 6 where a solution containing AQ4N or a salt of AQ4N is treated with activated charcoal.
- 20 8. A process for the preparation of compound AQ4N of formula (2)

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that includes the reaction step:

wherein the said reaction step is conducted in a stirrable solvent at a temperature not exceeding 200°C.

- 9. A process according to claim 8 wherein the solvent is tetramethylene sulfone.
- 10. A process according to claims 8 or 9 where the crude
 10 compound DDA of formula (6) is treated by slurrying several
 times with aqueous hydrochloric acid.
- 11. A process according to any one of claims 8 to 10 where the crude compound DDA of formula (6) is treated by adding a 15 chelating agent.
 - 12. A process for the preparation of compound AQ4N of formula (2)

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according to claim 1 which includes the reaction step:

wherein the reaction solution of the said reaction step is treated with an ammonium hydroxide and brine solution cooled to 0°C .

CLAIMS

1. A process for the preparation of compound AQ4N of formula (2):

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or a salt or solvate thereof wherein the said process includes the reaction step:

where compound AQ4 of formula (1) is oxidised to compound

- 10 AQ4N of formula (2) with an oxidising agent at a reaction temperature not exceeding 10°C, where the oxidising agent is a peracid or salt of a peracid, and where the oxidising agent is added at a temperature not exceeding 0°C.
- 15 2. A process according to claim 1 where the oxidising agent is magnesium monoperoxyphthalate.

- 3. A process according to either claim 1 or claim 2 where the reaction is conducted at a temperature not exceeding 0°C.
- 4. A process according to any one of claims 1 to 3 where
 the reaction solvent is 1,2-propanediol, dichloromethane or
 an aliphatic alkyl alcohol.
 - 5. A process according to any one of claims 1 to 4 for the preparation of a salt of AQ4N, where the salt of AQ4N, or a solvate thereof, is prepared by reaction of compound AQ4N of formula (2) with a solution of hydrogen chloride.
- 6. A process according to any one of claims 1 to 5 where a solution containing AQ4N or a salt of AQ4N is treated with activated charcoal.
 - 7. A process for the preparation of compound AQ4N of formula (2)

20 that includes the reaction step:

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wherein the said reaction step is conducted in a stirrable solvent at a temperature not exceeding 200°C.

(4)

- 8. A process according to claim 7 wherein the solvent is tetramethylene sulfone.
- A process according to claims 7 or 8 where the crude compound DDA of formula (6) is treated by slurrying several
 times with aqueous hydrochloric acid.
 - 10. A process according to any one of claims 7 to 9 where the crude compound DDA of formula (6) is treated by adding a chelating agent.

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11. A process for the preparation of compound AQ4N of formula (2)

according to claim 1 which includes the reaction step:

wherein the reaction solution of the said reaction step is treated with an ammonium hydroxide and brine solution cooled to $0\,^{\circ}\text{C}$.